

Assignment 1

$$\begin{aligned} 1a) \text{ Bohr radius } a_0 &= \frac{h^2}{m k q^2} \\ &= \frac{m_{\text{elect}}}{m_{\text{muon}}} \frac{h^2}{m_{\text{elect}} k q^2} \\ &= \frac{1}{207} (0.53 \text{ \AA}) \\ &= 2.6 \times 10^{-3} \text{ \AA} \\ a_0 &= 2.6 \times 10^{-13} \text{ m} \end{aligned}$$

$$\begin{aligned} b) \text{ Rydberg Energy } E_R &= \frac{m k^2 q^4}{2 h^2} \\ &= \frac{m_{\text{muon}}}{m_{\text{elect}}} \frac{m_{\text{elect}} k^2 q^4}{2 h^2} \\ &= 207 \times 13.6 \text{ eV} \\ &= 2.8 \text{ keV} \end{aligned}$$

$$\begin{aligned} 2) \text{ Energy of emitted photon } h\nu &= E_{\text{Ryd}} \left[\frac{1}{1^2} - \frac{1}{4^2} \right] \\ &= 13.6 \text{ eV} \left(1 - \frac{1}{16} \right) \\ &= 12.75 \text{ eV} \end{aligned}$$

$$\begin{aligned} \text{Photon Momentum} &= \frac{12.75 \times 1.6 \times 10^{-19} \text{ J}}{3 \times 10^8 \text{ m/sec}} \\ &= 6.8 \times 10^{-27} \text{ kg } \frac{\text{m}}{\text{sec}} \end{aligned}$$

Conservation of Momentum implies atom acquires equal & opposite momentum.

$$\therefore \vec{p}_{\text{atom}} = -\vec{p}_{\text{photon}}$$

$$\begin{aligned} \therefore \text{atom recoil energy } E_{\text{Rec}} &= \frac{p_{\text{atom}}^2}{2m_{\text{atom}}} \\ &= \frac{(6.8 \times 10^{-27})^2}{2 \times 1.67 \times 10^{-28}} \text{ J} \\ &= 1.5 \times 10^{-27} \text{ J} \\ &= 9.3 \times 10^{-9} \text{ eV.} \end{aligned}$$

$$3) \quad E_H = -\frac{E_{\text{Ryd}}}{n^2}$$

One can show $E_{\text{He}^+} = -z^2 \frac{E_{\text{Ryd}}}{n^2}$ where $z = \text{nuclear charge}$

\therefore if $n_{\text{He}^+} = 2n_H \Rightarrow$ same energy level.

eg. H transition $2 \rightarrow 1$ has same energy as

He^+ transition $4 \rightarrow 2$.

$$4) \quad E = -\vec{\mu} \cdot \vec{B}$$

$$\text{For } \vec{\mu} \text{ opposite } \vec{B} \Rightarrow E = \mu B.$$

$$a) \quad E = \mu_B B.$$

$$= \frac{e\hbar}{2m_e} \cdot B_{\text{Earth}}$$

$$= \frac{1.6 \times 10^{-19} \times \frac{6.63 \times 10^{-34}}{2\pi}}{2 \times 9.11 \times 10^{-31}} \cdot 0.5 \times 10^{-4} \text{ Tesla.}$$

$$= 4.6 \times 10^{-28} \text{ J}$$

$$= 2.4 \times 10^{-9} \text{ eV}$$

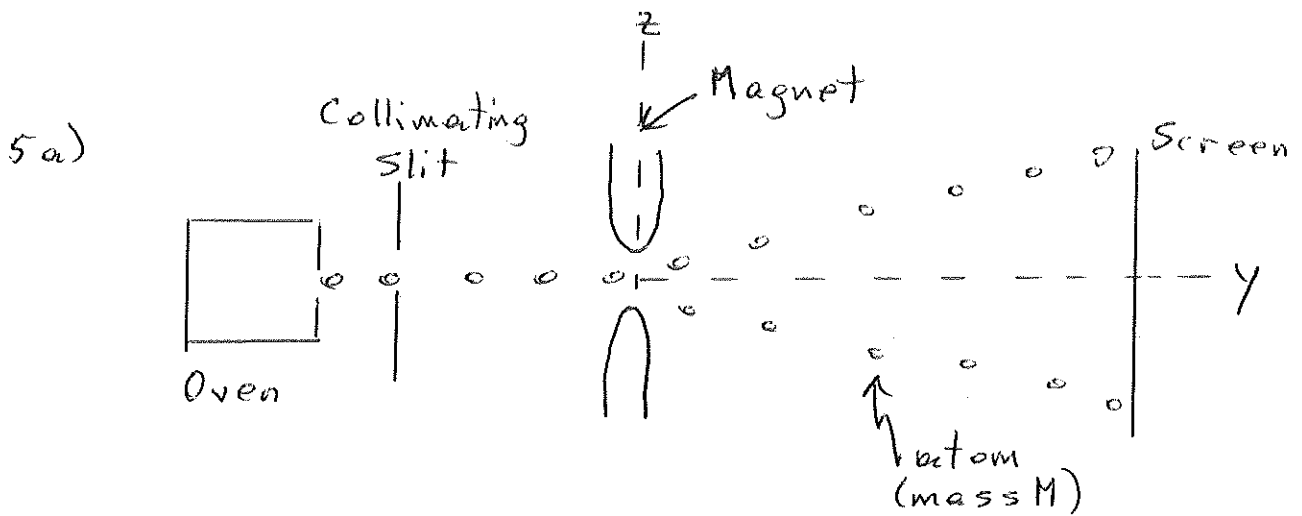
$$b) \quad E = \mu_{\text{Nuc}} \cdot B$$

$$= \frac{e\hbar}{2m_{\text{prot}}} \cdot B$$

$$= \frac{m_{\text{elect}}}{m_{\text{prot}}} \cdot \mu_B B.$$

$$= \frac{1}{1830} \cdot 2.4 \times 10^{-9}$$

$$= 1.3 \times 10^{-12} \text{ eV.}$$



While traversing magnet, atom experiences acceleration:

$$a_z = \frac{1}{M} \mu_B \frac{dB}{dz}$$

Magnet has length $l_1 \Rightarrow$ atom acquires vertical speed

$$\begin{aligned} v_z &= a_z t_1 \\ &= a_z \frac{l_1}{v_y} \end{aligned}$$

When atom leaves magnet, one can show deflection on screen is given by

$$\begin{aligned} z &= v_z t_2 \\ &= v_z \frac{l_2}{v_y} \quad \text{where } l_2 \text{ is distance from magnet to screen.} \end{aligned}$$

Note: Here we ignored vertical deflection in magnet which you can show is negligible.

$$\therefore \text{deflection } z = \frac{1}{M} \mu_B \frac{dB}{dz} \frac{l_1}{v_y} \frac{l_2}{v_y}$$

$$= \mu_B \frac{dB}{dz} \frac{l_1 l_2}{M v_y^2}$$

Stern & Gerlach used Ag atoms.

$$\therefore z = 9.3 \times 10^{-24} \frac{\text{J}}{\text{Tesla}} \cdot \frac{1 \text{ Tesla}}{.01 \text{ m}} \frac{(0.05 \text{ m})(1 \text{ m})}{108 \times 1.67 \times 10^{-27} \text{ kg} \cdot \left(300 \frac{\text{m}}{\text{sec}}\right)^2}$$

$$= 2.9 \times 10^{-3} \text{ m.}$$

$$\approx 3 \text{ mm.}$$

$v_{\text{thermal}} \approx \sqrt{\text{sound speed}}$